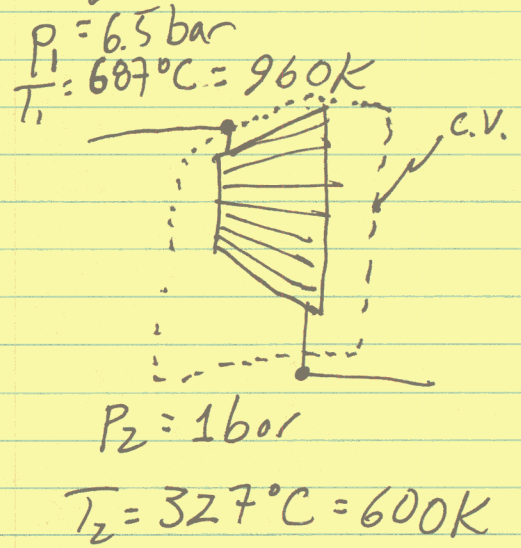


6.88 Given: Air, S.S., insulated turbine, ideal gas, $\Delta KE = \Delta PE = 0$

(a) $\frac{\dot{W}}{\dot{m}} = ?$



$$\frac{dE}{dt} = \dot{Q} - \dot{W} + \dot{m}_i \left(h_i + \frac{V_i^2}{2} + gz_i \right) - \dot{m}_e \left(h_e + \frac{V_e^2}{2} + gz_e \right)$$

S.S. insulated $\Delta KE = 0$ $\Delta PE = 0$

$$\frac{\dot{W}}{\dot{m}} = h_i - h_e$$

From table A-22

$$h_i = 1000.55 \text{ kJ/kg}$$

$$h_e = 607.02 \text{ kJ/kg}$$

$$\frac{\dot{W}}{\dot{m}} = 393.53 \text{ kJ/kg}$$

(b) Determine if this is reversible, irreversible or impossible

$$\frac{dS}{dt} = \sum_j \frac{\dot{Q}_j}{T_j} + \sum_i \dot{m}_i \Delta s_i - \sum_e \dot{m}_e \Delta s_e + \dot{\sigma}$$

S.S. \dot{m}_w

$$\frac{\dot{\sigma}}{\dot{m}} = \Delta s_e - \Delta s_i = \Delta A$$

For an ideal gas $\Delta A = \Delta^\circ(T_2) - \Delta^\circ(T_1) - R \ln\left(\frac{P_2}{P_1}\right)$

$$= 2.409 \text{ kJ/kg}\cdot\text{K} - 2.9213 \text{ kJ/kg}\cdot\text{K} - .287 \frac{\text{kJ}}{\text{kg}\cdot\text{K}} \ln\left(\frac{1 \text{ bar}}{6.5 \text{ bar}}\right)$$

$$\frac{\dot{\sigma}}{\dot{m}} = \Delta s_e - \Delta s_i = 0.0249 \text{ kJ/kg}\cdot\text{K} > 0$$

So \Rightarrow IRREVERSIBLE