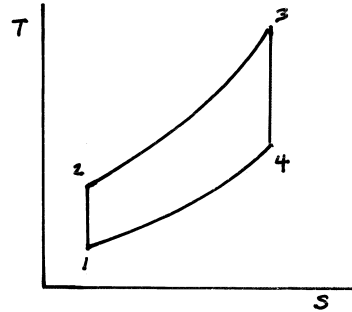
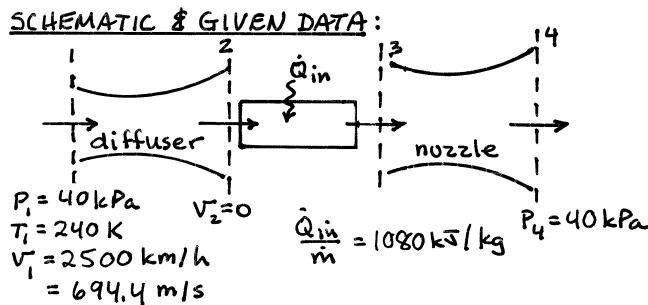


PROBLEM 9.82*

KNOWN: Air enters the diffuser of a ramjet with known conditions. The heat addition per unit mass of air flowing is specified.

FIND: Using air-standard analysis, determine (a) the pressure at the diffuser exit and (b) the velocity at the nozzle exit.

SCHEMATIC & GIVEN DATA:



ASSUMPTIONS: (1) Each component is analyzed as a control volume at steady state. (2) The combustion process is modeled as a constant pressure heat addition. (3) The diffuser and nozzle operate isentropically. (4) Neglect kinetic energy at locations 2 & 3 and neglect potential energy throughout. (5) The working fluid is air modeled as an ideal gas.

ANALYSIS: (a) For the diffuser; $0 = (h_1 + \frac{V_1^2}{2}) - h_2$

$$\text{Thus } h_2 = h_1 + \frac{V_1^2}{2} = 240.02 \frac{\text{kJ}}{\text{kg}} + \frac{(694.4)^2}{2} \frac{\text{m}^2}{\text{s}^2} \left| \frac{1 \text{ N}}{1 \text{ kg} \cdot \text{m/s}^2} \right| \left| \frac{1 \text{ kJ}}{10^3 \text{ N} \cdot \text{m}} \right|$$

$$= 481.1 \text{ kJ/kg}$$

Interpolating in Table A-22; $T_2 \approx 479 \text{ K}$ and $P_{r2} = 7.197$

Therefore $P_2 = \left(\frac{P_{r2}}{P_{r1}} \right) P_1 = \left(\frac{7.197}{0.6355} \right) (40 \text{ kPa}) = 453.0 \text{ kPa}$

(b) State 3 is fixed by considering the heat addition process

$$0 = \dot{Q}_{in} + \dot{m} (h_2 - h_3)$$

or $h_3 = \frac{\dot{Q}_{in}}{\dot{m}} + h_2 = 1080 + 481.1 = 1561.1 \text{ kJ/kg}$

Again, from Table A-22; $T_3 \approx 1438 \text{ K}$, $P_{r3} = 504.0$

For isentropic expansion through the nozzle

$$P_{r4} = (P_4/P_3) P_{r3} = (40/453.0)(504.0) = 44.50 \Rightarrow h_4 = 805.78 \frac{\text{kJ}}{\text{kg}}$$

The velocity is obtained from; $0 = h_3 - (h_4 + \frac{V_4^2}{2})$

or $V_4 = \sqrt{2(h_3 - h_4)}$

$$= \sqrt{2(1561.1 - 805.78) \frac{\text{kJ}}{\text{kg}} \left| \frac{10^3 \text{ N} \cdot \text{m}}{1 \text{ kJ}} \right| \left| \frac{1 \text{ kg} \cdot \text{m/s}^2}{1 \text{ N}} \right|}$$

$$= 1229 \text{ m/s}$$